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Polymorphism for a novel phosphoramidate; NMR and X-ray crystallography

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Abstract

New phosphoramidates with formula $3\text{-NC}_5H_4C(0)\text{NHP}(0)XY$ (X=Y=CI (1), X=Y=NH-C(CH₃)₃ (2a,2b), X=Y=N(C₄H₉)₂ (3), X=CI, Y=N(C₂H₅)₂ (4) were synthesized and characterized by IR, $^1\text{H-}$, $^{13}\text{C-}$, $^{31}\text{P-NMR}$ spectroscopy and CHN elemental analysis. Surprisingly, the reaction of compound 2a with LaCl₃, 7H₂O in 3:1 M ratio leads to a polymorph of this compound (2b). NMR spectra indicate that $^2\text{J}(\text{PNH}_{amide})$ in 2b (7.0 Hz) is very much greater than in 2a (4.1 Hz), while $\delta(^{31}\text{P})$ values are identical for both of them. In IR spectra, $\upsilon(\text{P=O})$ is weaker but $\upsilon(\text{C=O})$ is stronger in 2a than in 2b. The structures of 2a, 2b were determined by X-ray crystallography. These compounds form centrosymmetric dimers via two intermolecular P=O..... H-N hydrogen bonds. Strong intermolecular N-H... N, N-H... O and weak C-H... O hydrogen bonds lead to a three-dimensional polymeric cluster in the 2a while intermolecular strong N-H..... N and weak C-H..... O hydrogen bonds form a two-dimensional polymeric chain in 2b. © 2010 Springer Science+Business Media, LLC.

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