## Structural Chemistry

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# Polymorphism for a novel phosphoramidate; NMR and X-ray crystallography 

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#### Abstract

New phosphoramidates with formula $3-\mathrm{NC}_{5} \mathrm{H}_{4} \mathrm{C}(\mathrm{O}) \mathrm{NHP}(\mathrm{O}) \mathrm{XY}\left(\mathrm{X}=\mathrm{Y}=\mathrm{Cl}(1), \mathrm{X}=\mathrm{Y}=\mathrm{NH}-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right.$ (2a,2b), $X=Y=N\left(C_{4} H_{9}\right)_{2}(3), X=\mathrm{Cl}, \mathrm{Y}=\mathrm{N}\left(\mathrm{C}_{2} \mathrm{H}_{5}\right)_{2}$ (4) were synthesized and characterized by IR, ${ }^{1} \mathrm{H}-,{ }^{13} \mathrm{C}-,{ }^{31} \mathrm{P}-\mathrm{NMR}$ spectroscopy and CHN elemental analysis. Surprisingly, the reaction of compound 2a with $\mathrm{LaCl}_{3}, 7 \mathrm{H}_{2} \mathrm{O}$ in 3:1 M ratio leads to a polymorph of this compound (2b). NMR spectra indicate that ${ }^{2}{ }^{2}\left(\mathrm{PNH}_{\text {amide }}\right)$ in $2 \mathrm{~b}(7.0 \mathrm{~Hz})$ is very much greater than in 2 a ( 4.1 Hz ), while $\delta\left({ }^{31} \mathrm{P}\right.$ ) values are identical for both of them. In IR spectra, $u(P=O)$ is weaker but $\mathrm{u}(\mathrm{C}=0)$ is stronger in 2 a than in 2 b . The structures of $2 \mathrm{a}, 2 \mathrm{~b}$ were determined by X -ray crystallography. These compounds form centrosymmetric dimers via two intermolecular $\mathrm{P}=\mathrm{O} \ldots \ldots \mathrm{H}-\mathrm{N}$ hydrogen bonds. Strong intermolecular $\mathrm{N}-\mathrm{H} . . . \mathrm{N}, \mathrm{N}-\mathrm{H} . . \mathrm{O}$ and weak C-H... O hydrogen bonds lead to a three-dimensional polymeric cluster in the 2a while intermolecular strong $\mathrm{N}-\mathrm{H} . . . . . \mathrm{N}$ and weak $\mathrm{C}-\mathrm{H} . . . .$. O hydrogen bonds form a two-dimensional polymeric chain in 2b. © 2010 Springer Science+Business Media, LLC.


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